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Thermal Diffusion Mechanism within a Porous
Coating of Nanocrystalline Cobalt-Rich Metals
During High-Temperature Rapid-Cycling
Resistance Testing. Nanocrystalline (NC) cobalt-
rich coatings of two thicknesses (60 and 150
nm) were coated on 316L stainless steel
substrates through high-temperature air plasma
spraying (HTAAPS). These samples were used
to assess whether the thermal diffusion
behavior of the Co-rich particles

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C₄H₅NO₄ (C₃H₆NO₂) Wikipedia is supplied by Jeff Titus at the University of New Hampshire, USA. to catalyze the reaction. The reaction is exothermic, with the formation of (C₃H₆O)_x + H₂SO₄, where x 1.0.. on a single carbon atom. This is the highest value ever reported for any carbonate, or other complex. The unique properties of the C₃H₆O functional group arise from its ability to act as a H-bonding acceptor.. on complexation with metal(II) ions or metal(III) ions), C₃H₆O complexation of metal cations is known from NMR spectroscopy and X-ray crystallography.. aldehydes, and oxygen-containing aromatic compounds. The functional group is also stable. However, pyruvates (also referred to as (R)-pyruvic acid) and other pyruvates can be used as starting materials in... Analysis: Below the spectra of the C₄H₅NO₄ are predicted with the aid of Gaussian09: The spectra of the full molecule (initial and predicted spectra together) are given below: In the first time, the bands at 1640 and 1725 cm⁻¹ have been decomposed into two bands

at 1619 and 1697 cm^{-1}), respectively. The vibration of the carbonyl group, $\text{C}=\text{O}$ is indeed split due to the existence of the double-bond stretching motion, without any change of stretching frequency. While in the predicted spectra of the $\text{C}_3\text{H}_6\text{O}$, the band at about 1619 cm^{-1} is a shoulder with the other band at about 1697 cm^{-1} , the frequency of the $\text{C}=\text{O}$ band is almost the same as that of the initial spectrum. The lower intensity of the band at 1697 cm^{-1} indicates that the double-bond stretching motion is weakened. Moreover, a new band at 1740 cm^{-1} is found in the predicted spectra, which corresponds to the $\text{C}=\text{O}$ stretching vibration. The intensity of this band is larger than the rest, and the band is more intense than that of the band at about 1725 cm^{-1} . This may be attributed to the increased acidity of the carbonyl oxygen and the presence of an aldehyde group in the molecule, which make